

10/6/17, 546

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***** STN. Columbus *****

2004-06-24 18:43:35.000 24 JUN 2004

=> fil reg
COST IN U.S. DOLLARS
FILL ESTIMATED COST

FILE 'REGISTRY' ENTERED AT 19:43:33 ON 24 JUN 2004
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Property values tagged with IC are from the ZIC/VINITI data file provided by InfoChem.

STRUCTURE FILE UPDATES: 23 JUN 2004 HIGHEST RN 698346-19-9
DICTIONARY FILE UPDATES: 23 JUN 2004 HIGHEST RN 698346-19-9

TSCA INFORMATION NOW CURRENT THROUGH JANUARY 6, 2004

Please note that search-term pricing does apply when conducting SmartSELECT searches.

Crossover limits have been increased. See HELP CROSSOVER for details.

Experimental and calculated property data are now available. For more information enter HELP PROP at an arrow prompt in the file or refer to the file summary sheet on the web at:
<http://www.cas.org/ONLINE/DBSS/registryss.html>

```
=> s (salbutamol sulfate)/cn
L1           1 (SALBUTAMOL SULFATE) /CN
```

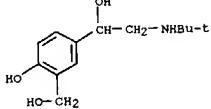
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=> d 11
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L1 ANSWER 1 OF 1 REGISTRY COPYRIGHT 2004 ACS on STN
 RN 51022-70-9 REGISTRY
 CN 1,3-benzenedimethanol, α 1-[(1,1-dimethylethyl)amino]methyl]-4-hydroxy-, sulfate (2:1) (salt) (9CI) (CA INDEX NAME)
 OTHER NAMES:
 CN (t)-Salbutamol sulfate
 CN Aerolin
 CN Airet
 CN Albuterol hemisulfate
 CN Albuterol sulfate
 CN Almotex
 CN Anebran
 CN Asmadil
 CN Asmalin
 CN Asmanil
 CN Asmasal
 CN Asmatol
 CN Asmavent
 CN Asmavent
 CN Asmidon
 CN Asmol Uni-Dose
 CN Asthalin
 CN Broncho-Spray
 CN Broncovaleas
 CN Bronter
 CN Bugonol
 CN Butamol
 CN Buto-Ahma
 CN Butotol
 CN Buventol
 CN Cetasin
 CN Cobutolin
 CN Dilatamol
 CN di-Salbutamol sulfate
 CN Ecovent
 CN Farcolin
 CN Grafalim
 CN Instavent
 CN Libretin
 CN Loftan
 CN Medolin
 CN Mozal
 CN Novosalmol
 CN NSC 289928
 CN Parasma
 CN Proventil
 CN Respax
 CN Salbetol
 CN Salbron
 CN Salbulin
 CN Salbumol
 CN Salbusian
 CN Salbutalan
 CN Salbutamol hemisulfate
 CN Salbutamol sulfate
 ADDITIONAL NAMES NOT AVAILABLE IN THIS FORMAT - Use FCN, FIDE, or ALL for
 DISPLAY
 DR 36519-31-0
 MF C13 H21 N O3 . 1/2 H2 O4 S
 CI COM

L1 ANSWER 1 OF 1 REGISTRY COPYRIGHT 2004 ACS on STN (Continued)
 LC STN Files: ADISINSIGHT, ADISNEWS, AGRICOLA, ANABSTR, BEILSTEIN*,
 BIOBUSINESS, BIOSIS, BIOTECHNO, CA, CAPLUS, CBMB, CEN, CHEMCATS,
 CHEMLIST, CIN, CSCHEM, DIGENES, EMBASE, IFICDB, IFIPAT, IFIUDB,
 IMSPATENT, IPA, MRCK*, MSDS-OHS, PHAR, PROMT, PROUSDDR, PS, RTECS*,
 SYNTHLINE, TOXCENTER, USAN, USPAT2, USPATFULL
 (*File contains numerically searchable property data)
 Other Sources: EINECS**
 (**Enter CHEMIST File for up-to-date regulatory information)
 DT.CA Caplus document type: Conference; Journal; Patent
 RL.P Roles from patents: BIOL (Biological study); MSC (Miscellaneous); PREP
 (Preparation); PROC (Process); PRP (Properties); RACT (Reactant or
 reagent); USES (Uses)
 RLD.P Roles for non-specific derivatives from patents: BIOL (biological
 study); USES (Uses)
 RL.NP Roles from non-patents: ANST (Analytical study); BIOL (Biological
 study); OCCU (Occurrence); PREP (Preparation); PROC (Process); PRP
 (Properties); RACT (Reactant or reagent); USES (Uses)
 RLD.NP Roles for non-specific derivatives from non-patents: BIOL (Biological
 study); USES (Uses)

CM 1

CRN 18559-94-9
 CMF C13 H21 N O3



CM 2

CRN 7664-93-9
 CMF H2 O4 S



587 REFERENCES IN FILE CA (1907 TO DATE)
 2 REFERENCES TO NON-SPECIFIC DERIVATIVES IN FILE CA
 592 REFERENCES IN FILE CAPLUS (1907 TO DATE)

=> fil caplus
COST IN U.S. DOLLARS
FULL ESTIMATED COST

SINCE FILE
ENTRY
7.04
TOTAL
SESSION
7.25

FILE 'CAPLUS' ENTERED AT 19:45:02 ON 24 JUN 2004
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FILE COVERS 1907 - 24 Jun 2004 VOL 140 ISS 26
FILE LAST UPDATED: 23 Jun 2004 (20040623/ED)

This file contains CAS Registry Numbers for easy and accurate substance identification.

=> s 51022-70-9/rn
592 51022-70-9
2 51022-70-9D
L2 590 51022-70-9/RN
(51022-70-9 (NOTL) 51022-70-9D)

=> s micro?
L3 2056097 MICRO?

=> s l2 and l3
L4 168 L2 AND L3

=> s ?milled
L5 31891 ?MILLED

=> s mill?
L6 245105 MILL?

=> s l3 or l5 or l6
L7 2270835 L3 OR L5 OR L6

=> s l2 and l7
L8 175 L2 AND L7

=> s vapor or gas
461479 VAPOR
68434 VAPORS
502025 VAPOR
(VAPOR OR VAPORS)
1363498 GAS
470088 GASES

1532221 GAS
(GAS OR GASES)
L9 1902692 VAPOR OR GAS
75% OF LIMIT FOR TOTAL ANSWERS REACHED

=> s vapor
461479 VAPOR
68434 VAPORS
L10 502025 VAPOR
(VAPOR OR VAPORS)

=> s gas
1363498 GAS
470088 GASES

SYSTEM LIMITS EXCEEDED - SEARCH ENDED

The search profile you entered was too complex or gave too many answers. Simplify or subdivide the query and try again. If you have exceeded the answer limit, enter DELETE HISTORY at an arrow prompt (=>) to remove all previous answers sets and begin at L1. Use the SAVE command to store any important profiles or answer sets before using DELETE HISTORY.

=> s l8 and l10
L11 10 L8 AND L10

=> d l11 1-10 abs ibib

L11 ANSWER 1 OF 10 CAPLUS COPYRIGHT 2004 ACS on STN
 AB The invention relates to a process for providing a stable crystalline form of a fine-milled salbutamol sulfate, which can be produced, stored and used while maintaining the aerodynamic properties required for inhalation. The process comprises the steps of micronizing salbutamol sulfate into a particle size required for inhalation, conditioning the salbutamol sulfate by treatment with a water-containing vapor, and drying the substance. The relative humidity was kept at $\geq 55\%$ so that the product recrystd. in 24 h. The stability of the product was dependent on the methods of conditioning.

ACCESSION NUMBER: 2004:60281 CAPLUS
 DOCUMENT NUMBER: 140199655
 TITLE: Process for Providing a stable crystalline form of salbutamol
 INVENTOR(S): Brodka-Pfeiffer, Katharina; Graas, Peter; Haesler, Heibert; Thieme, Herbert; Langguth, Peter
 PATENT ASSIGNEE(S): Boehringer Ingelheim Pharma GmbH & Co. KG, Germany
 SOURCE: PCT Int. Appl., 21 PP.
 CODEN: PIXXD2
 DOCUMENT TYPE: Patent
 LANGUAGE: English
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

| PATENT NO. | KIND | DATE | APPLICATION NO. | DATE |
|---|------|----------|-----------------|------------|
| WO 2004006884 | A2 | 20040122 | WO 2003-EP6787 | 20030626 |
| WO 2004006884 | A3 | 20040401 | | |
| W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ | | | | |
| RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AT, BE, BG, CH, CY, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG | | | | |
| US 2004052734 | A1 | 20040318 | US 2003-617546 | 20030710 |
| PRIORITY APPLN. INFO.: | | | EP 2002-15701 | A 20020712 |
| | | | US 2002-408375P | P 20020905 |

L11 ANSWER 2 OF 10 CAPLUS COPYRIGHT 2004 ACS on STN
 AB Interactive mixts. were prepared containing 5% (weight/weight) salbutamol sulfate using various lactose carrier systems, including sieved fractions and blended mixts. of coarse and fine particles. The solid state and powder properties of the lactose carriers were examined by laser diffraction, differential scanning calorimetry, thermogravimetric anal., powder x-ray diffraction, vapor sorption gravimetry, rotating drum and atomic force microscopy. The in vitro aerosols deposition was determined using a twin-stage impinger with a Rotahaler at an airflow rate of 60 l/min. The fine particle fraction (FFP) of salbutamol sulfate was determined using a validated HPLC assay. All samples were highly crystalline with minimal moisture sorption and the major phase in all samples was α -lactose monohydrate. Significant differences in FFP were observed using the various carrier systems. FFP increased with decreasing carrier d50₀ ($r^2=0.919$) and increasing proportion of fine carrier particles (below 5 μm) ($r^2=0.841$). Carriers consisting of very large proportions of fine particles showed low FFP and did not fit the correlation. The presence of coarse carrier particle fractions was essential to achieve maximum FFP, which occurred when about 10% of fine carrier particles were present in the mixture. Dispersion characteristics may be related to the degree of drug aggregation on the carrier surface.

ACCESSION NUMBER: 2003:60467 CAPLUS
 DOCUMENT NUMBER: 139:265576
 TITLE: Influence of physico-chemical carrier properties on the in vitro aerosols deposition from interactive mixtures
 AUTHOR(S): Louey, Margaret D.; Razia, Sultana; Stewart, Peter J.
 CORPORATE SOURCE: Victorian College of Pharmacy, Department of Pharmaceutics, Monash University, Parkville, Vic, Australia
 SOURCE: International Journal of Pharmaceutics (2003), 252(1-2), 87-98
 CODEN: IJPHDE; ISSN: 0378-5173
 PUBLISHER: Elsevier Science B.V.
 DOCUMENT TYPE: Journal
 LANGUAGE: English
 REFERENCE COUNT: 25 THERE ARE 25 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L11 ANSWER 3 OF 10 CAPLUS COPYRIGHT 2004 ACS on STN
 AB This study monitored the effect of a series of structurally related surfactants on the crystallization of amorphous salbutamol sulfate. Amorphous salbutamol sulfate was prepared by spray drying from a solution in water and in the presence of various alkyl polyglycosides (APGs) at different concns. The particles were then analyzed using isothermal microcalorimetry and water vapor sorption (Dynamic Vapor Sorption, DVS) anal. combined with near-IR spectroscopy (DVS-NIR). Both isothermal microcalorimetry and DVS-NIR were able to detect the transition from the amorphous to the crystalline state. The presence of APG surfactants modified the shape of the crystallization peak obtained using isothermal microcalorimetry. The gravimetric study combined with NIR revealed that while the crystallization was similar for the particles with or without surfactant, there was a great difference in the release of water from the newly formed crystal. In the presence of some of the surfactants tested, salbutamol sulfate released the water much faster than in the absence of surfactant. These results helped to explain the differences found in the isothermal microcalorimeter data. Differences were observed in the shapes of the NIR water peaks related to water due to the presence of the surfactant. In conclusion, the use of DVS combined with NIR has helped to analyze and understand the effect of APGs on the crystallization of amorphous salbutamol sulfate.

ACCESSION NUMBER: 2002:940635 CAPLUS
 DOCUMENT NUMBER: 139:154674
 TITLE: The effect of alkyl polyglycoside surfactants on the crystallization of spray-dried salbutamol sulphate: a gravimetric near-infrared spectroscopy study
 AUTHOR(S): Columbano, Angela; Buckton, Graham; Wikeley, Philip
 CORPORATE SOURCE: Dep. Pharmaceutics, Sch. Pharmacy, Univ. London, London, WC1N 1AX, UK
 SOURCE: PharmSci (online computer file) (2002), 4(3), No pp. given
 CODEN: PHARF; ISSN: 1522-1059
 URL: <http://www.aapspharmsci.org/scientificjournals/pharmsci/journal/pdf/pa040316.pdf>
 PUBLISHER: American Association of Pharmaceutical Scientists
 DOCUMENT TYPE: Journal; (online computer file)
 LANGUAGE: English
 REFERENCE COUNT: 9 THERE ARE 9 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L11 ANSWER 4 OF 10 CAPLUS COPYRIGHT 2004 ACS on STN
 AB The crystallization of amorphous salbutamol sulfate prepared by spray drying was monitored using a humidity controlled microbalance (Dynamic Vapor Sorption apparatus, Surface Measurement Systems) combined with a near-IR probe. Amorphous salbutamol sulfate was prepared by spray drying from a solution in water. The particles were then analyzed by SEM, thermogravimetric anal., DSC, powder x-ray diffraction, isothermal microcalorimetry and water vapor sorption anal. combined with near-IR spectroscopy (NIR). Isothermal microcalorimetry and water vapor sorption combined with NIR spectroscopy were able to detect the transition from the amorphous to crystalline state. However, while the isothermal microcalorimeter showed only a classic crystallization exotherm when the material was exposed at 75% RH, the DVS-NIR results at the same humidity highlighted a more complex process. When exposed at 75% RH, the uptake of water was followed by crystallization that was detected using NIR. The expulsion of water after crystallization was very slow and at a constant rate whether the material was exposed to 75 or 0% RH. The NIR and DVS studies indicated that the material had crystallized very soon after exposure to high RH. The water that was expelled during crystallization was not displaced from the particles and remained associated with the particles for many days. The use of gravimetric anal. together with NIR spectroscopy provided valuable information on the dynamics of the crystallization of salbutamol sulfate. The retention of water within recently crystallized salbutamol is potentially important to the behavior of dosage forms containing the amorphous (or partially amorphous) form of this drug.

ACCESSION NUMBER: 2002:278840 CAPLUS
 DOCUMENT NUMBER: 138:78246
 TITLE: A study of the crystallization of amorphous salbutamol sulphate using water vapor sorption and near infrared spectroscopy
 AUTHOR(S): Columbano, Angela; Buckton, Graham; Wikeley, Philip
 CORPORATE SOURCE: School of Pharmacy, Department of Pharmaceutics, University of London, London, WC1N 1AX, UK
 SOURCE: International Journal of Pharmaceutics (2002), 237(1-2), 171-178
 CODEN: IJPHDE; ISSN: 0378-5173
 PUBLISHER: Elsevier Science B.V.
 DOCUMENT TYPE: Journal
 LANGUAGE: English
 REFERENCE COUNT: 9 THERE ARE 9 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L11 ANSWER 5 OF 10 CAPLUS COPYRIGHT 2004 ACS on STN
 AB Particles of an amino acid such as leucine may be formed from an amino acid vapor, for example by aerosol condensation, or by spray drying. The amino acid particles have a bulk d. of not more than 0.1 g/cm³ or have a mass median aerodynamic diameter of not more than 10 μ mm

or are in the form of flakes having a thickness of not more than 100 μ mm. The inclusion of the particles of amino acid in powder for use in dry powder inhalers has been found to improve the respirable fraction of the active material in the powder. Ground L-leucine particles were suspended from a fluidized bed by a flow of air and carried in a gas flow into the tube furnace, which was at a temperature ranging from 150-300° and sublimed. The vapor emitted from the furnace was mixed with cool air giving a cloud of condensed particles that were subsequently collected in a cyclone and membrane filter. The bulk d. of the powder

was 0.04 g/cm³. A mixture of salbutamol sulfate and L leucine was prepared. The powder flow and handling performance of the salbutamol powder

was significantly improved, with minimal adhesion to glass walls compared with the milled leucine mixture.

ACCESSION NUMBER: 2000:401625 CAPLUS
 DOCUMENT NUMBER: 133:48937

TITLE: Pharmaceutical powders comprising particles of an amino acid

INVENTOR(S): Gorderton, David; Morton, David Alexander Vodden; Lucas, Paul

PATENT ASSIGNEE(S): Vectura Limited, UK
 SOURCE: PCT Int. Appl., 45 pp.

CODEN: PIIXXDZ
 DOCUMENT TYPE: Patent
 LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

| PATENT NO. | KIND | DATE | APPLICATION NO. | DATE |
|---|------|----------|---------------------------|----------|
| WO 2000033811 | A2 | 20000615 | WC 1999-GB4156 | 19991209 |
| WO 2000033811 | A3 | 20001012 | | |
| W: AE, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CR, CU, CZ, DE, DK, DM, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, US, UZ, VN, YU, ZA, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM, RW, GH, GM, KE, LS, KW, SD, SL, SZ, TZ, UG, ZW, AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, BF, BJ, CF, CG, CT, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG | | | | |
| BR 9916102 | A | 20010904 | BR 1999-16102 | 19991209 |
| EP 1137399 | A2 | 20011004 | EP 1999-958404 | 19991209 |
| EP 1137399 | BI | 20030514 | | |
| R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO | | | | |
| TR 200101591 | T2 | 20011121 | TR 2001-20010159119991209 | |
| JP 2002531487 | T2 | 20020924 | JP 2000-586305 | 19991209 |
| AT 240093 | E | 20030515 | AT 1999-958404 | 19991209 |
| NZ 511965 | A | 20030926 | NZ 1999-511965 | 19991209 |
| PT 1137399 | T | 20030930 | PT 1999-958404 | 19991209 |
| ES 2198973 | T3 | 20040201 | ES 1999-958404 | 19991209 |
| AU 770461 | B2 | 20040219 | AU 2000-15777 | 19991209 |

L11 ANSWER 5 OF 10 CAPLUS COPYRIGHT 2004 ACS on STN (Continued)
 NO 2001002825 A 20010608 NO 2001-2825 20010608
 PRIORITY APPLN. INFO.: GB 1998-27145 A 19981209
 WO 1999-GB4156 W 19991209

L11 ANSWER 6 OF 10 CAPLUS COPYRIGHT 2004 ACS on STN
 AB There are described finely divided particles of pharmaceutical substance, wherein the substance when submitted to water vapor gives off heat of less than 1.2 J per g, processes for their production

and pharmaceutical formulations containing them. An example is given of salbutamol sulfate (25%) and lactose (75%) conditioned with water at relative humidity 55-65%, nonconditioned micronized substance mixture (5-8 J/g) and conditioned micronized mixture (<0.5 J/g).

ACCESSION NUMBER: 1999:133202 CAPLUS
 DOCUMENT NUMBER: 130:200925
 TITLE: Finely divided pharmaceutical particles for inhalation
 INVENTOR(S): Briggner, Lars-Erik; Bystrom, Katarina; Jakupovic, Edib; Trofast, Eva; Trofast, Jan
 PATENT ASSIGNEE(S): Astra Aktiebolag, Swed.
 SOURCE: U.S., 7 pp., Cont.-in-part of U.S. Ser. No. 459,660.
 DOCUMENT TYPE: Patent
 LANGUAGE: English
 FAMILY ACC. NUM. COUNT: 3
 PATENT INFORMATION:

| PATENT NO. | KIND | DATE | APPLICATION NO. | DATE |
|---|----------------|----------|-----------------|----------|
| US 5874063 | A | 19990223 | US 1996-606655 | 19960226 |
| AU 9215347 | AI | 19921117 | AU 1992-15347 | 19920324 |
| AU 662519 | B2 | 19950907 | | |
| EP 580648 | AI | 19940202 | EP 1992-907877 | 19920324 |
| EP 580648 | BI | 19960508 | | |
| R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, MC, NL, PT, SE | | | | |
| JP 06506454 | T2 | 19940721 | JP 1992-507195 | 19920324 |
| JP 3400999 | B2 | 20030428 | | |
| EP 680752 | A2 | 19951108 | EP 1995-111178 | 19920324 |
| EP 680752 | A3 | 19951122 | | |
| EP 680752 | B1 | 20011114 | | |
| R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, MC, NL, PT, SE | | | | |
| PL 168232 | B1 | 19960131 | PL 1992-301008 | 19920324 |
| RU 2112507 | C1 | 19980610 | RU 1993-58260 | 19920324 |
| SK 280310 | B6 | 19991108 | SK 1993-1088 | 19920324 |
| CZ 286936 | B6 | 20000816 | CZ 1993-2116 | 19920324 |
| JP 2003155228 | A2 | 20030527 | JP 2002-347368 | 19920324 |
| NO 9303575 | A | 19931006 | NO 1993-3575 | 19931006 |
| US 5709884 | A | 19980120 | US 1995-379471 | 19950130 |
| US 5637620 | A | 19970610 | US 1995-459660 | 19950602 |
| US 5562923 | A | 19961008 | US 1995-479494 | 19950607 |
| PRIORITY APPLN. INFO.: | SE 1991-1090 | | A 19910411 | |
| | SE 1993-2777 | | A 19930827 | |
| | US 1993-129204 | | B1 19931025 | |
| | US 1995-379471 | | B3 19950130 | |
| | US 1995-459660 | | A2 19950602 | |
| | US 1995-479494 | | A2 19950607 | |
| | SE 1996-141 | | A 19960116 | |
| | CS 1993-2116 | | A 19920324 | |
| | EP 1992-907877 | | A3 19920324 | |
| | JP 1992-507195 | | A3 19920324 | |
| | WO 1992-SE186 | | A 19920324 | |
| | WO 1994-SE780 | | W 19940825 | |

L11 ANSWER 6 OF 10 CAPLUS COPYRIGHT 2004 ACS on STN (Continued)
 RECORD. ALL CITATIONS AVAILABLE IN THE REFORMAT

L11 ANSWER 7 OF 10 CAPLUS COPYRIGHT 2004 ACS on STN
 AB Pharmaceutical powders are often milled to achieve the optimum particle size. These size reduction processes can introduce dislocations and/or defects onto particle surfaces affecting the overall crystallinity of the powder. If enough energy is imparted, amorphous regions on the particles surfaces may be produced. These amorphous regions have the propensity to absorb significant quantities of water. In this study, the effect of sorbed water on the phys. characteristics of albuterol sulfate was investigated. Phys. properties of this compound were studied in both micronized and unmicronized states using SEM, DSC, powder X-ray diffraction, solution microcalorimetry, laser diffraction particle size anal. and water vapor sorption anal. Subtle differences in crystallinity induced by air jet micronization were detected by several anal. methods. Amorphous to crystalline conversions were observed, the

Kinetic of which are found to be both temperature and relative humidity dependent. These expts. show the dynamic nature of micronized albuterol sulfate and acid in the determination of the actual phys.

state of this pharmaceutical powder.

ACCESSION NUMBER: 1995:556628 CAPLUS

DOCUMENT NUMBER: 122:298909

TITLE: Process-induced crystallinity changes in albuterol sulfate and its effect on powder physical stability

AUTHOR(S): Ward, Gary R.; Schultz, Robert K.

CORPORATE SOURCE: 3M Pharmaceuticals, St. Paul, MN, 55144-1000, USA

SOURCE: Pharmaceutical Research (1995), 12(5), 773-9

CODEN: PHREB; ISSN: 0724-8741

PUBLISHER: Plenum

DOCUMENT TYPE: Journal

LANGUAGE: English

L11 ANSWER 8 OF 10 CAPLUS COPYRIGHT 2004 ACS on STN
 AB The present invention relates to a process for providing a stable crystalline form to a fine-grained substance or a substance mixture, which can be produced, stored and used while maintaining the aerodynamic properties required for inhalation of such a substance or a substance mixture, by a) in case of a substance mixture, preparing a homogeneous mixture of the substances; b) micronizing, direct precipitating or diminishing by any conventional method the substance or substance mixture into a particle size required for inhalation, the particle size being less than 10 μ m; c) optionally preparing a homogeneous mixture of the desired substances when each substance has been introduced from stage b) as sep. fine-grained particles; d) conditioning said substance or substance mixture by treatment with a water containing vapor phase in a controlled fashion; and e) drying.

ACCESSION NUMBER: 1995:528673 CAPLUS

DOCUMENT NUMBER: 122:274076

TITLE: Process for conditioning substances

INVENTOR(S): Trofast, Eva Ann-Christin; Briggner, Lars-Erik

PATENT ASSIGNEE(S): Astra Aktiebolag, Swed.

SOURCE: PCT Int. Appl., 20 pp.

DOCUMENT TYPE: Patent

LANGUAGE: English

FAMILY ACC. NUM. COUNT: 3

PATENT INFORMATION:

| PATENT NO. | KIND | DATE | APPLICATION NO. | DATE |
|---|------|----------|------------------|----------|
| WO 9505805 | A1 | 19950302 | WO 1994-SE780 | 19940825 |
| W: AM, AT, AU, BB, BG, BR, BY, CA, CH, CN, CZ, DE, DK, EE, ES, FI, GB, GE, HU, JP, KE, KG, KP, KR, KZ, LK, LT, LU, LV, MD, MG, MN, MW, NL, NO, NZ, PL, PT, RO, RU, SD, SE, SI, SK, TV, TT, UA, US, UZ, VN | | | | |
| RW: KE, MW, SD, AT, BE, CH, DE, DK, ES, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, BF, BJ, CF, CG, CI, CM, GA, GN, ML, MR, NE, SN, TD, | | | | |
| TG | | | | |
| ZA 9405675 | A | 19960429 | ZA 1994-5675 | 19940825 |
| TW 427916 | B | 20010401 | TW 1994-83107152 | 19940804 |
| IL 110698 | A1 | 20021110 | IL 1994-110698 | 19940818 |
| CA 2170394 | AA | 19950302 | CA 1994-2170394 | 19940825 |
| AU 9476264 | A1 | 19950321 | AU 1994-76264 | 19940825 |
| AU 681186 | B2 | 19970821 | | |
| BR 9407320 | A | 19960416 | BR 1994-7320 | 19940825 |
| EP 717616 | A1 | 19960626 | EP 1994-926421 | 19940825 |
| EP 717616 | B1 | 20010321 | | |
| R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IE, IT, LU, MC, NL, PT, | | | | |
| SE | | | | |
| CN 1133004 | A | 19961009 | CN 1994-193793 | 19940825 |
| CN 1049333 | B | 20000216 | | |
| HU 74000 | A2 | 19961028 | HU 1996-447 | 19940825 |
| HU 217770 | B | 20000428 | | |
| JP 09501930 | T2 | 19970225 | JP 1994-507516 | 19940825 |
| JP 2978247 | B2 | 19991115 | | |
| PL 176749 | B1 | 19990730 | PL 1994-313142 | 19940825 |
| RU 2148992 | C1 | 20000520 | RU 1996-105935 | 19940825 |

L11 ANSWER 8 OF 10 CAPLUS COPYRIGHT 2004 ACS on STN (Continued)
 AT 199828 E 20010415 AT 1994-926421 19940825
 ES 2156158 T3 20010616 ES 1994-926421 19940825
 PT 717616 T 20010830 PT 1994-926421 19940825
 CZ 289018 B6 20011017 CZ 1996-544 19940825
 SK 283146 B6 20030304 SK 1996-234 19940825
 US 5709884 A 19980120 US 1995-379471 19950130
 NO 9600744 A 19960223 NO 1996-744 19960223
 FI 9600869 A 19960226 FI 1996-869 19960226
 CN 1195523 A 19981014 CN 1997-123049 19971126
 CN 1090019 B 20020904
 HK 1016493 A1 20030425 HK 1999-101600 19990414
 GR 3036106 T3 20010928 GR 2001-400955 20010621
 PRIORITY APPLN. INFO.: SE 1993-2777 A 19930827
 WO 1994-SE780 W 19940825

L11 ANSWER 9 OF 10 CAPLUS COPYRIGHT 2004 ACS on STN
 AB Isothermal microcalorimetry has been used to monitor the recrystn. of spray-dried salbutamol sulfate. The drug recrystallizes in water vapor, by a cooperative process. The cooperative nature demonstrates that the water must first absorb to saturate the entire powder bed before recrystn. occurs. Consequently, recrystn. is slower for low humidities, due to a slower arrival of water vapor. The data have been compared with previous data for recrystn. of spray-dried lactose. The heat change for the crystallization was significantly lower for salbutamol sulfate than for lactose. In terms of apparent enthalpy of crystallization, the large exothermic responses are indicative of the fact that the crystal form is the thermodynamically stable state. The salbutamol which had been recrystd. at the lower humidities showed that the process, while being rapid, was discontinuous. In each case, the exothermic recryst. was followed by an endothermic response for the expulsion of water as the amorphous region recrystd. There was a repeating sequence of crystallization, followed by water expulsion, followed by further recrystn. With each repeat of the cycle the responses decreased in size. This ability to follow crystallization in real time provides a novel insight into the process.

ACCESSION NUMBER: 1995:365096 CAPLUS
 DOCUMENT NUMBER: 122:165903
 TITLE: The use of isothermal microcalorimetry in the study of changes in crystallinity of spray-dried salbutamol sulfate
 AUTHOR(S): Buckton, Graham; Darcy, Patricia; Greenleaf, David; Holbrook, Paula
 CORPORATE SOURCE: Centre for Materials Science, The School of Pharmacy, University of London, 29-39 Brunswick Square, London, WC1N 1AX, UK
 SOURCE: International Journal of Pharmaceutics (1995), 116(1), 113-18
 CODEN: IJPHD; ISSN: 0378-5173
 PUBLISHER: Elsevier
 DOCUMENT TYPE: Journal
 LANGUAGE: English

L11 ANSWER 10 OF 10 CAPLUS COPYRIGHT 2004 ACS on STN
AB Extending the residence time of drugs delivered to the lungs as
inhalation
aerosols may results in sustained therapeutic drug levels and reduced
toxicity. Droplets were generated from 0.25 wt% disodium fluorescein
(DF), and 0.25 wt% albuterol sulfate solns. at a rate of 1 mL min⁻¹ using
a Turbotac jet nebulizer. These droplets were dried, concentrated and
mixed
with saturated lauric acid (LA) vapor at both temps. of
60-140°. The resulting coated particles were <5 µm in size as
estimated by inertial impaction and SEM. Powder composition, as
determined by gas
chromatog., ranged from ratios of 1.2:1 to 2.5:1, of LA:DF. Evidence of
coating of DF by LA was derived from IR spectroscopy and x-ray
microanal. Dissoln. studies performed on the coated particles in
phosphate buffer, pH 7.4, at 37° and quantified by UV spectroscopy,
showed that the half-time for dissoln. (t_{1/2}) increased from 4 min for
uncoated DF particles to 22-55 min for lauric acid coated DF particles,
depending on the coating thickness. The t_{1/2} for albuterol sulfate
particles increased from 2.5 min to 12.5 min for albuterol sulfate
particles coated with lauric acid at a bath temperature of 100°.
Inhalation studies performed on beagle dogs with DF particles coated with
lauric acid (bath temperature, 100°) indicated there was a shift and
broadening of the peak plasma concentration in comparison with aerosols
of DF
alone. The average absorption half-time increased from 4.7 min for
uncoated
DF particles to 11.5 min for lauric acid coated DF particles.
ACCESSION NUMBER: 1994:491637 CAPLUS
DOCUMENT NUMBER: 121:91637
TITLE: Controlled release from condensation coated
respirable
aerosol particles
AUTHOR(S): Pillai, R. S.; Yeates, D. B.; Miller, I. F.; Hickey,
A. J.
CORPORATE SOURCE: Dep. Chem. Eng., Univ. Illinois, Chicago, IL, 60680,
USA
SOURCE: Journal of Aerosol Science (1994), 25(3), 461-77
CODEN: JALSS7; ISSN: 0021-8502
DOCUMENT TYPE: Journal
LANGUAGE: English

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